

How to describe a solid:

- by its size (from big to small):

lump	chip flake	granule	grain filing speck	powder fine powder or coarse powder	particle
	flocculate				
to <i>grind</i> (grind – ground – ground) to grind to powder	to <i>break off</i> a chip to <i>chip off</i>		to <i>file</i> a piece of metal		

A solid can be dissolved in a liquid. Let's take common salt:

Salt is **dissolved** in water. The result of this process is a **solution**. This means that salt is **soluble** (adj.); it is a **solute**. The water, however, is a **solvent**.

How to describe a liquid:

Grade of purity	When looking through it	When something is dissolved in it		
Pure	Clear	saturated	Viscous	Volatile (adj)
Impure or contaminated	milky creamy murky and ?*	unsaturated	To measure its <i>viscosity</i> is part of physical chemistry	Volatility (n)

Two liquids can be mixed, that means that they are **miscible**, others cannot be mixed and are therefore **immiscible**.

* Does a state between liquid and solid exist? Perhaps one may consider certain saturated *suspensions* to have an undefined state of matter. *Plasma* as the 4th state of matter must also be mentioned in this context.